

Corrections

Excess Volumes of Binary Mixtures Containing Diisopropyl Ether + 1-Butanol or Diisopropyl Ether + Diethyl Ketone and Ethanol + Heptane at High Pressures. P. Ulbig,* M. Bubolz, C. Kornek, and S. Schulz, *J. Chem. Eng. Data* 1997, 42, 449–452.

The densities given in Table 2 for pure diisopropyl ether were incorrect. The correct Table 2 is given below.

Table 2. Densities of Pure Liquid Components at Different Temperatures and Pressures

T/K	component	$\rho/\text{g}\cdot\text{cm}^{-3}$			
		$p = 0.1$ MPa	$p = 20$ MPa	$p = 40$ MPa	$p = 60$ MPa
278.15	diisopropyl ether	0.7388	0.7581	0.7737	0.7869
	diethyl ketone	0.8300	0.8446	0.8570	0.8677
	1-butanol	0.8206	0.8331	0.8440	0.8539
283.15	methanol		0.8190	0.8324	0.8444
	methanol		0.8170	0.8313	0.8437
	(Tongfan, 1988)				
298.15	diisopropyl ether	0.7182	0.7397	0.7571	0.7719
	diethyl ketone	0.8037	0.8190	0.8328	0.8450
	1-butanol	0.8060	0.8193	0.8311	0.8418
	ethanol	0.7814	0.7975		0.8235
	heptane	0.6739	0.6906		0.7170
323.15	diisopropyl ether	0.6913	0.7167	0.7369	0.7532
	1-butanol	0.7854	0.8004	0.8133	0.8252

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Absorption of H₂S by an Aqueous Methyldiethanolamine Solution at 296 and 343 K. F. Pani, A. Gaunand, D. Richon,* R. Cadours, and C. Bouallou, *J. Chem. Eng. Data* 1997, 42, 865–870.

Tables A2–A25 of the Appendix were not made available at the time of publication. The Tables are now accessible.

Supporting Information Available:

Tables A2–A25 for the Appendix are available in electronic form via the Internet. Accessing information is given on any current masthead page.

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